

<b>Product Name</b>	Sodium Chloride — Laboratory Grade	<b>Cat. No.</b>	ASC-8051
<b>Molecular Formula</b>	NaCl	<b>MW (g/mol)</b>	58.44
<b>CAS Number</b>	7647-14-5	<b>EC Number</b>	231-598-3
<b>Grade</b>	Laboratory Grade / High Purity	<b>HS Code</b>	2849.10.00
<b>Synonyms</b>	Sodium Chloride; NaCl; Common Salt; Halite; Saline Salt; Table Salt (lab grade)		

## 1. Product Description

Sodium Chloride Laboratory Grade (ASC-8051) is a high-purity inorganic salt (NaCl, CAS 7647-14-5, assay  $\geq 99.5\%$ ) supplied as a white, free-flowing crystalline powder. It is one of the most widely used reagents in laboratory science — essential for the preparation of physiological saline, phosphate buffered saline (PBS), culture media, electrophoresis buffers, and osmolarity adjustment in biological and chemical experiments.

AuSaMicS ASC-8051 is manufactured and packed in Melbourne, Australia. Each batch is released with a full batch-specific Certificate of Analysis confirming compliance with all physicochemical specifications.

## 2. Physicochemical Specifications

Parameter	Specification	Method
Assay (NaCl)	$\geq 99.5\%$ (w/w)	Argentometric titration / potentiometric
<b>Appearance</b>	White crystalline powder or granules; free-flowing	Visual
pH (5% aq. solution, 25°C)	6.7 – 7.3	Potentiometric (calibrated electrode)
<b>Loss on Drying (105°C / 2 hr)</b>	$\leq 0.5\%$	Gravimetric
Insoluble Matter	$\leq 0.005\%$	Gravimetric filtration
<b>Sulfate (as SO<sub>4</sub><sup>2-</sup>)</b>	$\leq 0.005\%$	Turbidimetric (BaSO <sub>4</sub> )
Chlorate (as ClO <sub>3</sub> <sup>-</sup> )	$\leq 0.003\%$	Titrimetric
<b>Bromide (as Br<sup>-</sup>)</b>	$\leq 0.01\%$	ICP / titrimetric
Heavy Metals (as Pb)	$\leq 5$ ppm	ICP-OES / atomic absorption
<b>Iron (as Fe)</b>	$\leq 2$ ppm	Spectrophotometric
Calcium (as Ca)	$\leq 0.002\%$	Complexometric / ICP-OES
<b>Magnesium (as Mg)</b>	$\leq 0.001\%$	ICP-OES
Potassium (as K)	$\leq 0.005\%$	Flame photometry / ICP
<b>Nitrogen compounds (as N)</b>	$\leq 0.001\%$	Kjeldahl / spectrophotometric

## 3. Physical Data

Property	Value
Melting Point	801 degrees C
<b>Boiling Point</b>	1,465 degrees C
Density	2.165 g/cm <sup>3</sup> (at 25 degrees C)
<b>Solubility in Water</b>	357 g/L at 20 degrees C; 391 g/L at 100 degrees C
Refractive Index (nD)	1.5442
<b>Crystal System</b>	Cubic (face-centred cubic — rock salt structure)
Vapour Pressure	Negligible at room temperature

## 4. Typical Applications

Application Area	Typical Use
Buffer Preparation	Phosphate Buffered Saline (PBS), normal saline (0.9%), HEPES buffer, TRIS buffer, SSC buffer for hybridisation
<b>Culture Media</b>	Osmolarity adjustment; component in most microbiological culture media formulations
Molecular Biology	DNA/RNA extraction buffers, hybridisation solutions, cell lysis buffers, precipitation protocols
<b>Electrophoresis</b>	Running buffers for agarose and polyacrylamide gel electrophoresis
Osmolarity Control	Cell culture osmolarity adjustment; cryoprotection solutions
<b>General Chemistry</b>	Titration, standard solutions, solution preparation, gravimetric analysis

## 5. Common Laboratory Recipes

### Phosphate Buffered Saline (1x PBS, pH 7.4, per litre):

- NaCl (ASC-8051): 8.0 g
- KCl: 0.2 g
- Na<sub>2</sub>HPO<sub>4</sub>: 1.44 g
- KH<sub>2</sub>PO<sub>4</sub>: 0.24 g
- Dissolve in 800 mL distilled water. Adjust pH to 7.4 with HCl. Make up to 1 litre. Autoclave 121°C / 15 min.

### Normal Saline (0.9% NaCl, per litre):

- NaCl (ASC-8051): 9.0 g
- Dissolve in 1 litre distilled water. Autoclave 121°C / 15 min or filter-sterilise (0.2 µm).

## 6. Preparation Notes

1. Weighing: NaCl is mildly hygroscopic. Weigh quickly in a dry environment and close the stock container immediately after use.
2. Dissolution: Dissolve in distilled or deionised water with stirring at room temperature. NaCl dissolves readily — no heating required for standard concentrations (<350 g/L).
3. pH: NaCl solutions are nearly pH-neutral (pH 6.7-7.3). Adjust pH after all buffer components are dissolved and before final volume adjustment.
4. Sterilisation: Autoclave NaCl solutions at 121°C for 15 minutes. Alternatively, filter-sterilise through a 0.2 µm membrane for heat-sensitive applications.

## 7. Storage & Shelf Life

Store at 15-25°C in a tightly closed container, away from moisture.

NaCl is hygroscopic — reseal immediately after use. Use a desiccant in high-humidity storage areas.

Shelf life: Indefinite under proper dry storage conditions. Purity is not affected by age.

Do not expose to excessive humidity or direct moisture.

## 8. Literature References

5. Haynes WM (ed). CRC Handbook of Chemistry and Physics. 97th ed. CRC Press; 2016. Sodium Chloride physical and chemical data.
6. Sambrook J, Russell DW. Molecular Cloning: A Laboratory Manual. 3rd ed. Cold Spring Harbor Laboratory Press; 2001. PBS and saline buffer protocols.
7. ISO 6353-1:1982. Reagents for chemical analysis — Part 1: General testing methods. ISO, Geneva.
8. British Pharmacopoeia 2023. Sodium Chloride monograph. MHRA, London.

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